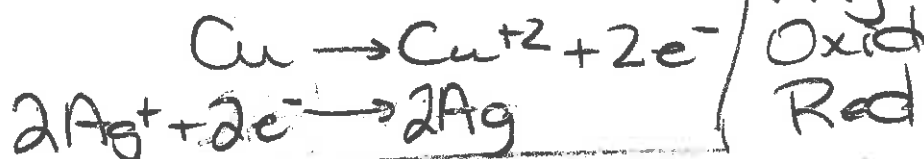
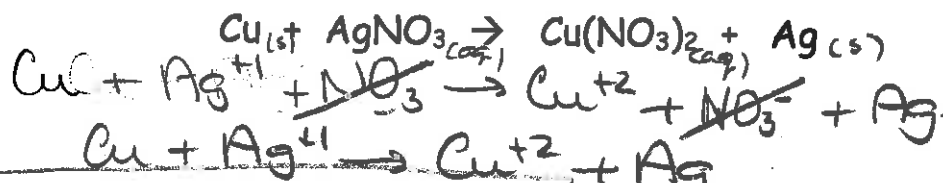
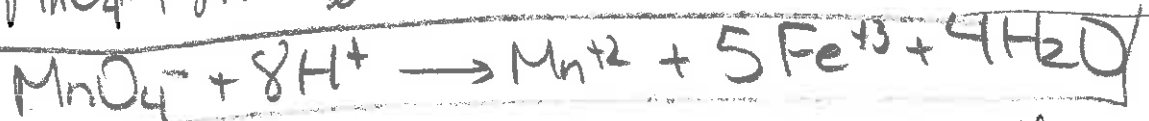
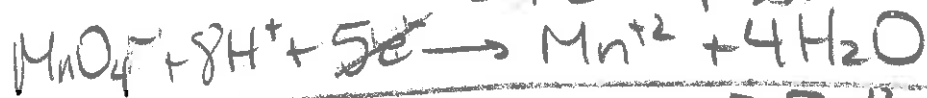
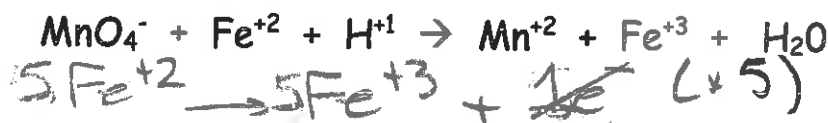


Writing Half Reactions



Balancing Using Half Reactions

1. Write the balanced half reactions.
2. Change the coefficients in the balanced half reactions so that the number of electrons produced in the oxidation half reaction equals the number of electrons consumed in the reduction half reaction.
3. Add the 2 half reactions to obtain the balanced equation!



* Compare to Ox # method * they are the same.

