## A. Significant Digits

- 1. Non-zero digits are always significant. 127.34 grams = 5 significant digits (s.d.)
- 2. All zeroes between nonzero digits are significant. 1205 m = 4 s.d.
- 3. All final zeroes to the right of a decimal point are significant. 21.50 grams = 4 s.d.
- 4. Zeroes used only for spacing the decimal are not significant. 0.0025 = 2 s.d.
- 5. Scientific notation is used to indicate if zeroes at the end of measurements are significant.
  - ➤ When adding & subtracting round off or drop digits until the answer has the same number of decimal places as the fewest being added or subtracted.
  - When multiplying & dividing round the final answer to the least number of significant digits.

## **B.** Scientific Notation

Expresses a number by writing it in the form  $a \times 10^n$  where a is  $\le 1$  but < 10 and the digits in the coefficient a are all significant.

## To change from standard form to scientific notation:

- 1. Place a decimal after the first non-zero digit.
- 2. Count the number of spaces you moved the decimal. This is the exponent of 10.
- 3. Look at the direction you moved your decimal:

Left = positive exponent (e.g. 10<sup>4</sup>)

Right = negative exponent (e.g.  $10^{-4}$ )

In other words, if the original number is larger than 1, the exponent is positive. If the original number is smaller than 1, the exponent is negative.

4. Write the numerical value in scientific notation.

## To change from scientific notation to standard form:

- 1. Move the decimal to the right for a positive exponent of 10.
- 2. Move the decimal to the left for a negative exponent of 10.
- 3. Write the value in standard form.

## Examples:

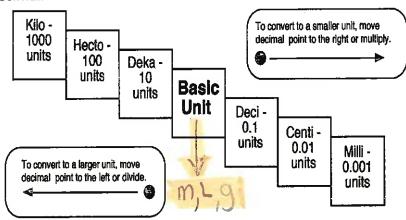
a) 1571 -> 4 sd b) 19001 -> 5 sd c) 3.680 -> 4 sd d) 0.03680 -> 4 sd e) 3.00×104 -> 3 sd 1.902×10-3 -> 4 sd 4) 13.2+13.92+ 14.111 = 40.2 9) 3.6×10.22×1.09 = 40 = 4.0×10'

10,700 →1.07×104 0.107 → 1.07×10-1

 $3.6 \times 10^3 \rightarrow 3600$  $3.6 \times 10^3 \rightarrow 0.036$ 

### C. Converting Units

To convert prefixes, such as when using the same base unit, simply move the decimal.



To convert derived units, remember that multiplying or dividing by 1 does not alter an equation.

### Examples:

Common Conversion Factors		Example Problems:		
Mass	1000 g = 1 kg	How many grams are in 32.5 kg? $30.5 \text{ kg} \times \frac{1000 \text{ g}}{1 \text{ kg}} = 30.500 \text{ g}$		
Length	1000 mm = 1 m 100 cm = 1 m 1000 m = 1 km	How many millimeters are in 8.45 km?  8.45 km/x 1000m/x 100cm/x 10mm 8450000 mm		
Time	60 s = 1 min 60 min = 1 hr 24 hrs = 1 day	How many seconds are in 2.5 days?  2.5 d x 24 x 3600s 216,000s		



#### D. Unit/Dimensional Analysis

Most physical quantities have dimensions that can be used to help solve problems.

	<u>Variable</u>	<u>Unit</u>
Х		m
V		m/s
t		s
а		m/s²

### E. Formula Manipulation

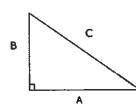
RULE #1: you can add, subtract, multiply and divide by anything, as long as you do the same thing to both sides of the equals sign.

RULE #2: to move or cancel a quantity or variable on one side of the equation, perform the "opposite" operation with it on both sides of the equation.

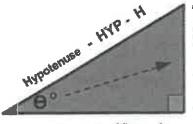
### F. Solving Problems - GRASP Method

- 1. Given state the known quantities
- 2. Required state what is unknown
- 3. Analyze determine the equation(s) to be used, check for like units, organize the steps to solve, draw a picture if needed
- 4. Solve solve for the unknown variable
- 5. Paraphrase make sure the answer makes sense and is in correct units and sig digits

# G. Trigonometry



 $A^2 + B^2 = C^2$ 



Name	Ratio	Expression
Sine	0/H	Sine
Cosine	A/H	Cose
Tangent	O/A	Tane

Adjacent - ADJ - A

Weu	se	"SOH-CA	H-TOA"	to help u:	s remen	nber the F	Ratios
		short for		= Oppos = Adjace			= O/H = A/H
		short for					= O/A

# Examples:

$$x = vt + \frac{1}{2} at^2$$

Show this equation is dimensionally

correct:
$$[m] = [m][S] + [m][S]^{a}$$

$$= [m] + [m]$$

$$[m] = [m]$$

$$a = \frac{v_f - v_i}{\Delta t}$$

Solve for v<sub>f</sub>:

A boat has a 30.0m long cable with anchor attached. If the ship is anchored at an angle of 39°, how deep is the water?

