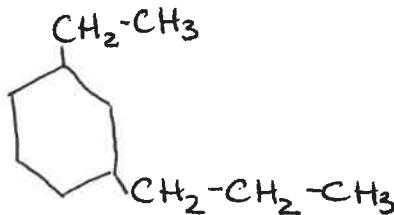


(1)

Pg 17 # 2e) 1,1 dimethylcyclohexane
f) isopropylcyclopentane

4d)



Pg 31 # 1 a) chlorobenzene

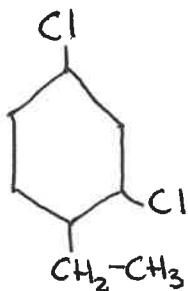
b) methylbenzene

c) phenylethene

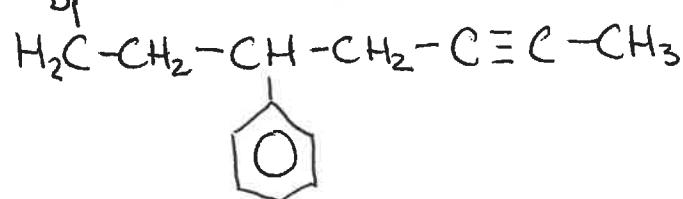
d) 1,3-dimethylbenzene

e) 1-bromo-2-pentylbenzene

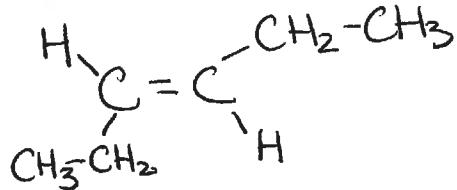
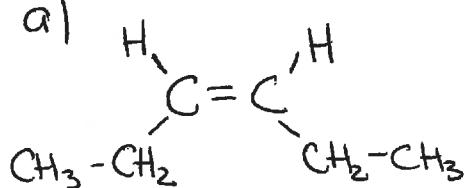
2 a)



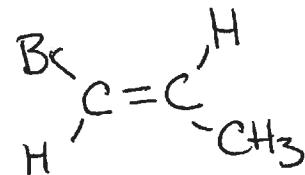
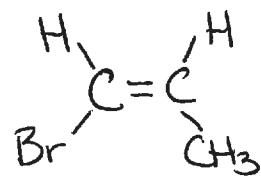
b)



Pg 23 # 1 a)



b)



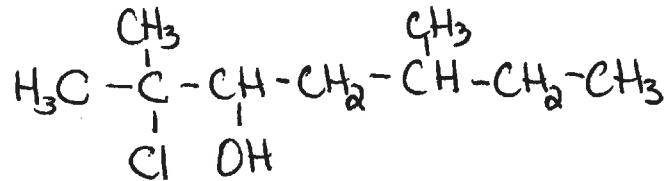
2 a) trans pent-2-ene

b) cis pent-2-ene

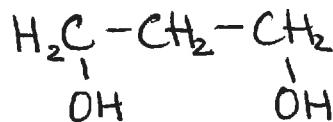
(2)

- Pg 34 #1 a) pentane 1,4-diol
 b) octan-4-ol
 c) benzene-1,3-diol

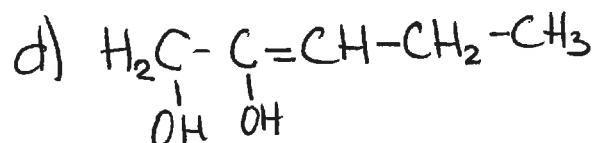
#2 a)



b)



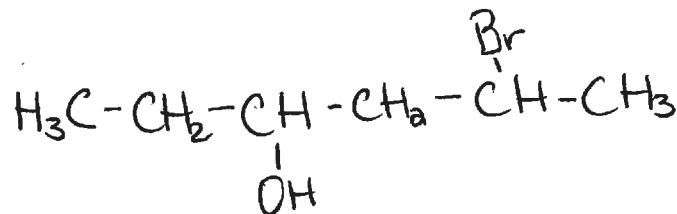
c)



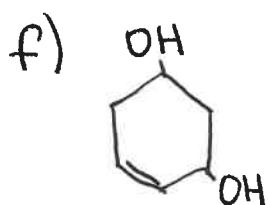
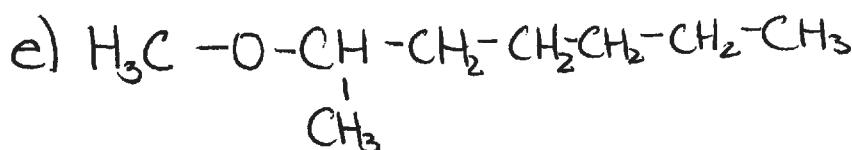
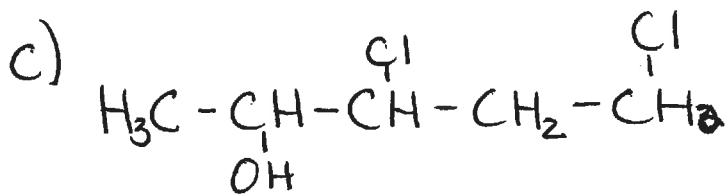
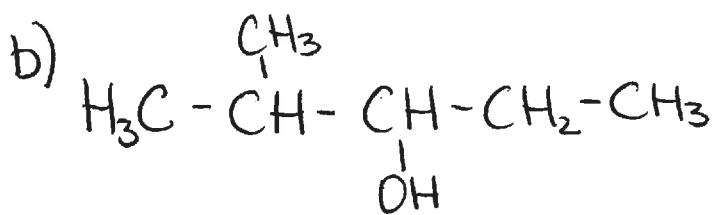
Pg 39 #1 a) heptane-2,3-diol

- b) 3-methylhexan-2-ol
 c) 4-ethyl(-5-methylheptan-3-ol)
 d) cyclopentane-1,3-diol
 e) benzene-1,2,4-triol
 f) 1-propoxybutane
 g) 1-ethoxypentane
 h) ethanethiol

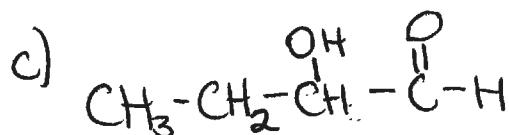
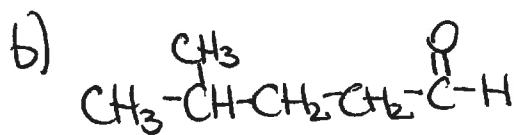
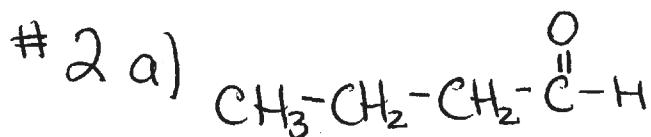
#2 a)



③

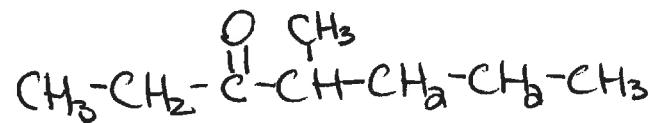


Pg 41 #1 a) pentanal b) 4-ethylhexanal
c) 4-chloropentanal

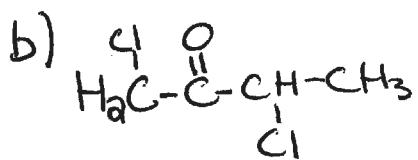


Pg 42 #1 a) butanone b) 2-methylhexan-3-one
c) cyclohexanone

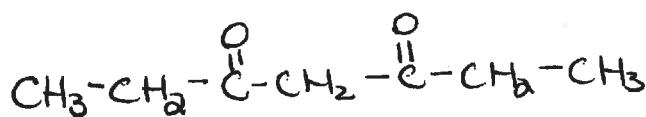
#2 a)



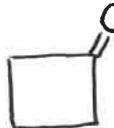
(4)



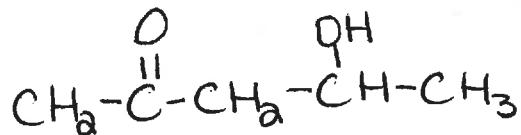
c)



d)



e)



Pg 46 #1

Name	Condensed structure	Line diagram or structural formula	Type of compound
heptanal	$\text{CH}_3\text{CH}_2\text{CH}_2\text{CH}_2\text{CH}_2\text{CH}_2\text{CHO}$		aldehyde
heptan-4-one	$\text{CH}_3\text{CH}_2\text{CH}_2\text{COCH}_2\text{CH}_2\text{CH}_3$		ketone
pentan-3-one	$\text{CH}_3\text{CH}_2\text{COCH}_2\text{CH}_3$		ketone
1-chlorobutan-2-one	$\text{CH}_2\text{ClCOCH}_2\text{CH}_3$		ketone
3-methylpentanal	$\text{CH}_3\text{CH}_2\text{CH}(\text{CH}_3)\text{CH}_2\text{CHO}$		aldehyde
2-methylbutanal	$\text{CH}_3\text{CH}_2\text{CH}(\text{CH}_3)\text{CHO}$		aldehyde

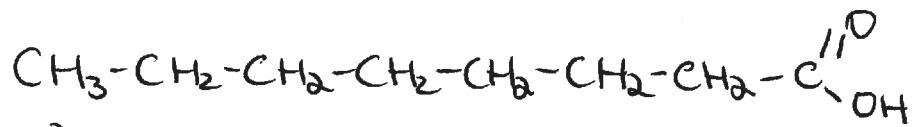
(5)

Pg 48 #1 a) decanoic acid

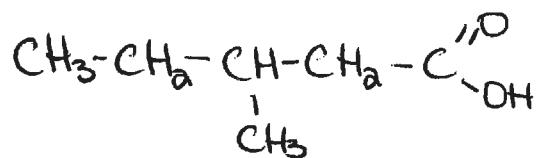
b) butanoic acid

c) 3-methylheptanoic acid

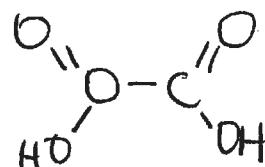
#2 a)



b)



c)



Pg 50 #1

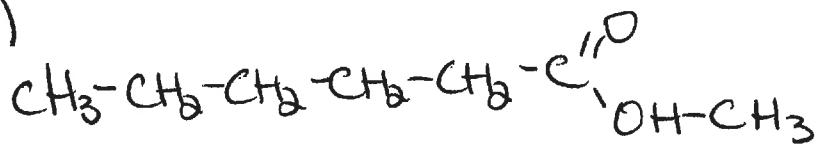
a) ethyl pentanoate

b) methyl decanoate

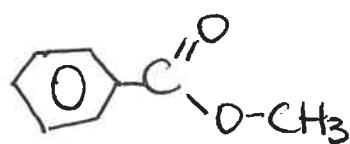
c) butyl methanoate

d) propyl benzoate

#2 a)



b)

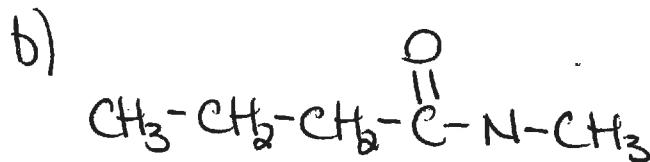
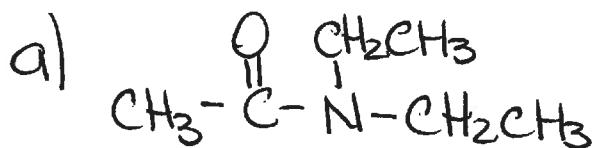


(6)

Pg 58 # 1

- a) butan-2-amine
- b) N,N-diethylbutan-1-amine
- c) N-methyl-N-propylpropan-1-amine
- d) Octan-3-amine
- e) N-ethylbutan-2-amine

Pg 60 # 2



Pg 62 # 2

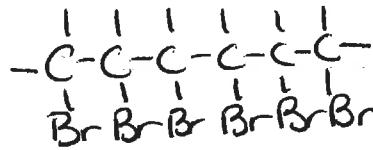
Table 2

Name	Condensed structure	Line diagram or structural formula	Type of compound
N,N-dimethylbutanamide	$\text{CH}_3\text{CH}_2\text{CH}_2\text{CON}(\text{CH}_3)_2$		amide
propan-1-amine	$\text{CH}_3\text{CH}_2\text{CH}_2\text{NH}_2$		amine
N-ethylpentanamide	$\text{CH}_3\text{CH}_2\text{CH}_2\text{CH}_2\text{CONHCH}_3$		amide
N-methylpentan-1-amine	$\text{CH}_3\text{CH}_2\text{CH}_2\text{CH}_2\text{CH}_2\text{NHCH}_3$		amine
N-methylbutan-1-amine	$\text{CH}_3\text{CH}_2\text{CH}_2\text{CH}_2\text{NHCH}_3$		amine
N-propylbutan-1-amine	$\text{CH}_3\text{CH}_2\text{CH}_2\text{CH}_2\text{NHCH}_2\text{CH}_2\text{CH}_3$		amine

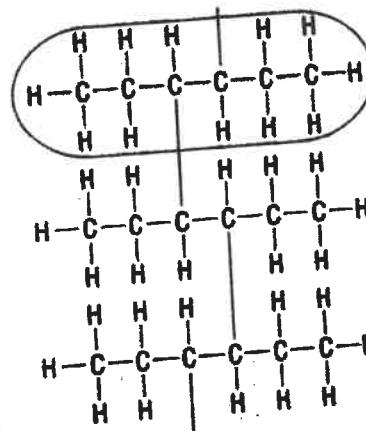
Pg 87 #1

(1)

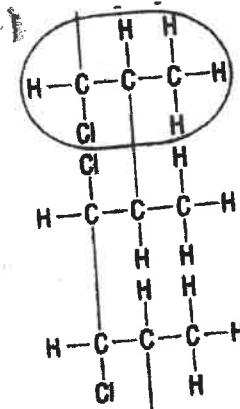
a) Polydibromoethene



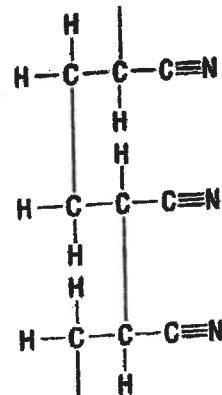
b) polyhex-3-ene



c) poly-1-chloropropene



#2 Cyanooctene

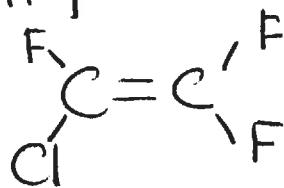


#3 Bromoethene



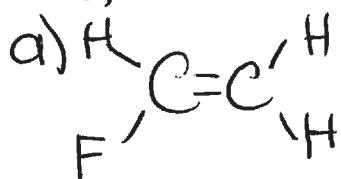
Pg 93 #1

(8)



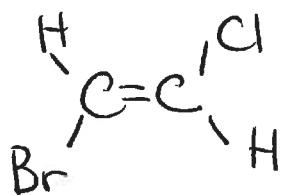
chlorotrifluoroethene

#2



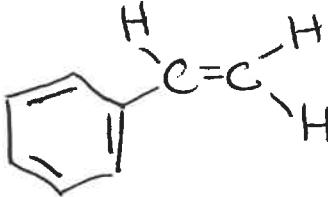
fluoroethene

b)



trans-1-bromo-2-chloroethene

c)



phenylethene